

Chemistry

Collaborative Learning



Burning Dominoes

Developed by Rose Elgar and Stuart Scott

Theme: Chemical reactions Age range: 11+

Context:

This dominoes activity is designed to consolidate the spelling and meanings of scientific vocabulary. We have left some of the cards blank for you to add your own. The dominoes can also be adapted by recording onto the cards terms in different languages - either for EAL learners or MFL.

Preparation:

Print out the A4 sheets.

If possible, laminate or cover the dominoes pages with 'clear pvc self adhesive book cover film'. Then cut out the squares to make the cards . The blank pages should not be covered so that students can write on them.

HOW TO PLAY:

Play in groups of two or three players.

Deal out all the dominoes. Whoever has the start card goes first and places it face up in the middle of the circle. Move clockwise around the circle, matching the terms with their meaning, if you do not have the correct card, move onto the next player.

Continue working round the group until one player has finished.

The winner is the first player to have no dominoes left.

Learning Outcomes:

Students will be able to identify the different types of chemical reactions and their meanings. If you have found further learning outcomes please share them by emailing collaborate@mantralingua.com.

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(You can purchase a professionally printed, sound-enabled version of this activity with pre-cut cards from www.mantralingua.com. Students can use TalkingPEN to record and re-record aurally onto the pages and cards. Recordings can be saved and used for assessment, or shared with other classes and schools via "ShareLINK".)



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a material that burns easily

flammable

oxygen

the gas needed for something to burn

oxide

chemical made when something reacts with oxygen

chemical change / reaction when a new substance is made that cannot easily be changed to what it was a substance that can be burned to produce energy

fuel

joules

the units used to measure energy

limewater

a liquid that turns milky when in contact with carbon dioxide gas

oxidation

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the chemical reaction that happens when fuel burns

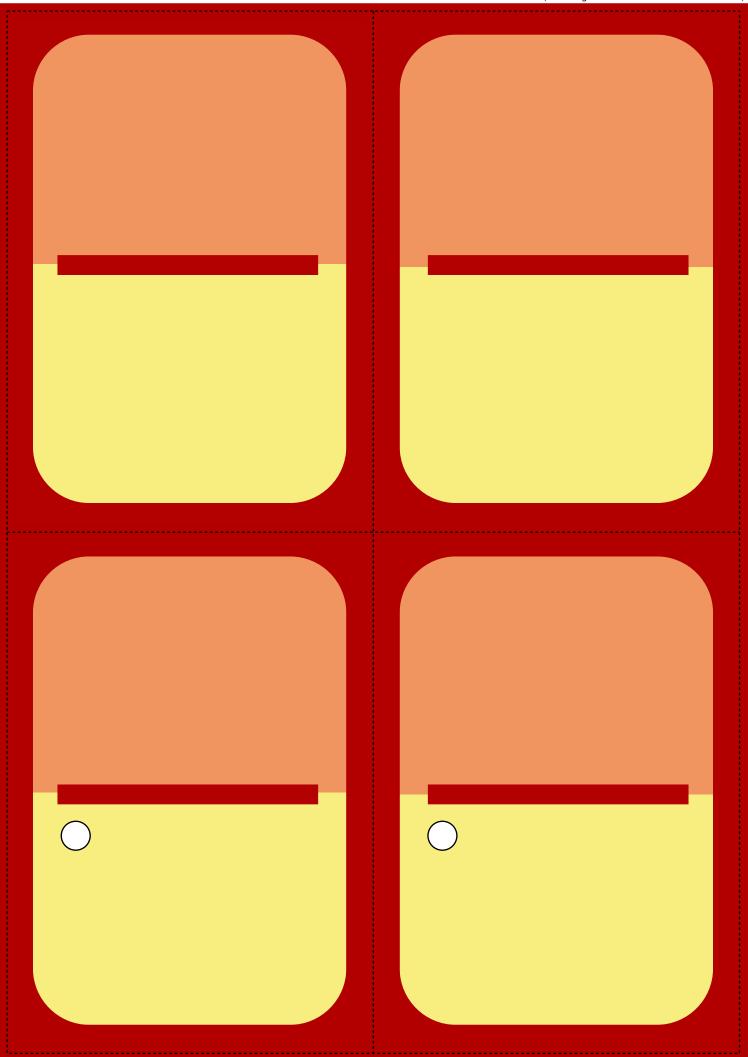
shows what happens in chemical reaction

equation

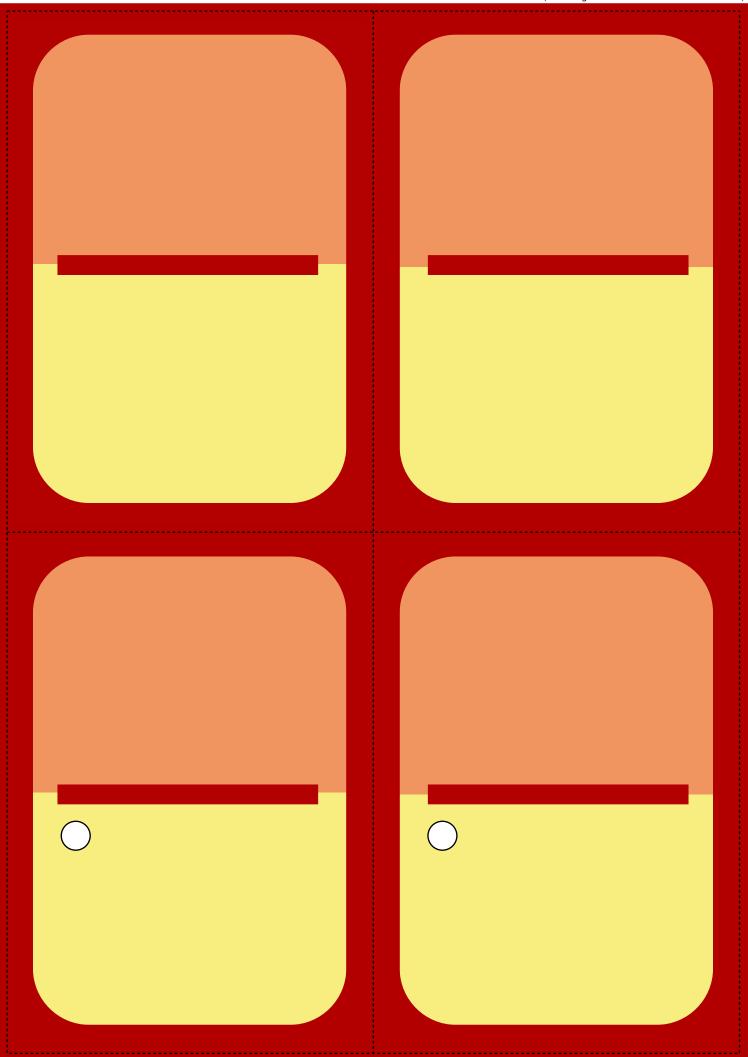
hydro-carbons

fuels that contain hydrogen and carbon only

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flammable	a material that burns easily
oxygen	the gas needed for something to burn
oxide	chemical made when something reacts with oxygen
chemical change/ reaction	when a new substance is made that cannot easily be changed back to what it was
fuel	a substance that can be burned to produce energy
joules	the units used to measure energy
limewater	liquid that turns milky when in contact with carbon dioxide gas
oxidation	the chemical reaction that happens when a fuel burns
equation	show what happens in a chemical reaction
hydro - carbons	fuels that contain hydrogen and carbon only